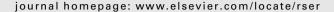


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# Renewable and Sustainable Energy Reviews





# Assessment of nitrogen fertilization for the CO<sub>2</sub> balance during the production of poplar and rye

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# ABSTRACT

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# Contents

1.	Introd	ductionduction	1454
2.	Meth	ods	1454
	2.1.	System boundary	1454
	2.2.	Study area	1454
	2.3.	Field design	1454
	2.4.	Direct N₂O emissions	1455
	2.5.	Indirect N <sub>2</sub> O emissions	1455
	2.6.	Harvesting of annual and perennial energy crops	1455
	2.7.	Statistical analyses	1455
3.	Resul	ts	1455
	3.1.	Biomass yield	1455
	3.2.	Soil N <sub>2</sub> O emissions	1455
	3.3.	$\mathrm{NH_3}$ and $\mathrm{NO_x}$ volatilization.	1456
	3.4.	Leaching	
	3.5.	CO <sub>2</sub> equivalents from the manufacturing of calcium ammonium nitrate (CAN)	1457
	3.6.	${ m CO_2}$ balance of the production of energy crops.	1457
4.	Discu	ssion	1458
	4.1.	Mineral N fertilizer	1458
	4.2.	Energy and C balance	1458
	4.3.	Assessment of energy crops	1458
5.	Concl	lusions	1459
	Ackno	owledgements	1459
	Refer	ences	1459

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#### 1. Introduction

In order to fulfill the requirements of the Kyoto Protocol and face climate change, the production of biomass has been intensified worldwide and has a great potential, which is far away to be fully used yet [1,2]. Energy crop cultivation makes it possible to remove CO<sub>2</sub>, and thermal conversion of these crops can act as a substitute for fossil fuels. A positive effect on CO<sub>2</sub> mitigation can be expected, however, this might be reduced by other greenhouse gases, which are released during the production and consumption of energy crops [3-6]. Emissions may occur either directly during agricultural operations such as fertilizing, ploughing and harvesting, or indirectly as a consequence of nitrate leaching, or during production and transport of fertilizers and pesticides. This study is not designed to employ a full life cycle assessment (LCA), which reflects a complete evaluation of all greenhouse gas sources and socio-economic aspects as emphasized by Hanegraaf et al. [7]. The focus of this study is on the use of mineral nitrogen fertilizer and its effect on the CO<sub>2</sub> balance of the production of one annual energy crop (rye) and one perennial energy crop (poplar) on a loamy sandy soil in Northeast Germany.

According to Wegener et al. [8], about 113 million tons of CO<sub>2</sub> equivalents are emitted from the German agricultural sector (Fig. 1). This amount does not include emissions linked with the production of mineral fertilizers. One-third (31.5%) of the agricultural sector derives from agricultural soils in the form of nitrous oxide (N2O), a greenhouse gas, which is very important due to its global warming potential, 298 times higher than that of CO<sub>2</sub>. Thus the CO<sub>2</sub> fixing capacity of renewable energy crops may be constrained. Agronomic practices such as tillage, harvesting and fertilizer applications can significantly affect the production and consumption of N<sub>2</sub>O due to alteration of soil physical and chemical characteristics, and biochemical activities. Soil cultivation and precipitation affect the soil air exchange rate and thus influence microbial aerobic and anaerobic processes such as nitrification and denitrification [9-11]. Both processes may be stimulated after the application of nitrogen fertilizer, as reported by Freney [12].

Even the production of mineral nitrogen fertilizers results in considerable N<sub>2</sub>O emissions besides the emission of CO<sub>2</sub> during this energy-consuming process [13,14]. In Western Europe 1.8% of total CO<sub>2</sub> and N<sub>2</sub>O emissions and 0.9% of total energy consumption derive from the fertilizer industry [15]. Consequently, the emission of greenhouse gases through this pre-chain of biomass production comprises an important component of agricultural LCA [16]. It has been reported that the production of short rotation coppices such as poplar in particular needs only a low supply of nitrogen fertilizer [17,18]. Thus it may be hypothesized that poplar has a greater potential for reducing CO<sub>2</sub> emissions than the annual crop rye. The overall goal is to obtain more verified information about the ecological drawbacks of nitrogen fertilization and to find out at

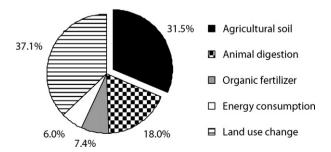


Fig. 1. Sources of greenhouse gases expressed as  $CO_2$  equivalents from agriculture in Germany according to Wegener et al. [8]. Greenhouse gas emissions from agricultural soils were the main subject of this study.

what fertilizer rate the highest net CO<sub>2</sub> sequestration can be expected from the production of rye and poplar.

#### 2. Methods

# 2.1. System boundary

This study focused on the growing period of energy crops and the utilization of nitrogen fertilizer in the field. From this on-farm stage, the main components included in the  $\rm CO_2$  balance are both, the carbon fixation by above-ground crop biomass and direct and indirect greenhouse gas emissions from the soil. Carbon sequestration by below-ground biomass and by soil was not included in this study.

Among the pre-farm stages, only the production of mineral N fertilizer is taken into consideration, since it is known as a highly energy-consuming process [19,20]. In Germany, 1.8 million tons of mineral fertilizer nitrogen was sold in 2007/2008. This amount includes 0.81 million tons of calcium ammonium nitrate (CAN) nitrogen [21]. Since CAN played the main role among all nitrogen fertilizers applied in German agriculture and also in other European countries during the last ten years, this type of mineral nitrogen fertilizer has been evaluated in terms of CO<sub>2</sub> equivalents related to its manufacturing process [22].

Greenhouse gas emissions arising during the transport of raw materials and their products have been reported as playing a minor role of between 1% and 3% of total emissions during biomass production [23], and have therefore been neglected. Biomass yields were obtained after harvesting; however harvesting and post-harvest processes themselves have not been considered in this  ${\rm CO_2}$  balance.

# 2.2. Study area

The study was performed from February 1999 to December 2007 on an experimental field at the Leibniz-Institute for Agricultural Engineering (ATB) in Potsdam, Germany (52°26′N,  $13^{\circ}00'E$ ). The topsoil (0–30 cm) can be characterized as loamy sand with a carbon content of up to 0.8% and pH 6.0 (Table 1). Deeper layers are relatively rich in organic matter, which must be due to former applications of lake sediments and other organic enrichments during a period of fruit-growing in the 1980s and before. The mean air temperature and precipitation during the study period were  $10.1\,^{\circ}\text{C}$  and  $580\,\text{mm}\,\text{yr}^{-1}$ .

# 2.3. Field design

The experimental field with a total area of 4.1 ha was parcelled out for the cultivation trials in 1994 [24]. In our study, a  $\mathrm{CO}_2$  balance was drawn up for one perennial and one annual crop. The perennial crop was poplar (*Populus maximowiczii*  $\times$  *P. nigra*), a short rotation crop with a four-year rotation period (study period 1999–2007), and the annual crop was rye (*Secale cereale* L.). Only the six years (1999, 2001, 2003, 2005, 2006 and 2007) when rye was planted at the same site were considered in case of rye. Both crops were arranged in

**Table 1** Soil characteristics with grain sizes from Scholz [46] and  $C_{\text{tot}}$  and  $N_{\text{tot}}$  analyses from December 2005.

Soil horizon	Depth (cm)	Grain size (%)		$C_{\text{Tot}} (\text{g kg}^{-1})$	$N_{\mathrm{Tot}} (\mathrm{gkg}^{-1})$	
		Sand	Silt	Clay		
Ap	0-30	77.9	15.9	6.2	7.6	1.04
Ah	30-60	75.7	18.3	6.0	5.8	0.75
Bt	60-82	62.0	20.6	17.4	5.8	0.49
Cv	>82	62.1	25.7	12.2		

rectangles each with an area of  $104~\text{m} \times 24~\text{m}$ , which were subdivided into four plots. Three times between April and May, nitrogen fertilizer (CAN) was applied to the plots with a total input of 75 N kg ha $^{-1}$  yr $^{-1}$  and 150 N kg ha $^{-1}$  yr $^{-1}$ . Further phosphorus and potassium fertilizers were applied every second year in the form of Kamex-superphosphate in the case of the 150 kg N plot and in the form of wood ash on the 75 kg N plot. An additional plot with no fertilizer applications was arranged as a control. No pesticides were applied throughout the study.

# 2.4. Direct N<sub>2</sub>O emissions

Direct  $N_2O$  emissions from the soil were measured using closed cylindrical chambers placed on a collar with a volume of 0.064 m<sup>3</sup> (upper diameter 0.39 m, lower diameter 0.51 m, height 0.40 m). Measurements were performed four times a week in the morning during the whole years. The chambers were located within the stand of poplar (row distance 2.25 m) and rye (row distance 0.15 m). Gases were sampled in 125 ml evacuated glass vials at the beginning and the end of incubation after 60 min. During the incubation period the increase of  $N_2O$  emissions had been proved for linearity. Gas samples were analyzed using a gas chromatograph according to Loftfield et al. [25] at the same day.

The fertilizer-induced  $N_2O$  emission rate was calculated for each rate of fertilization by taking the difference between the mean values of the fertilized and the non-fertilized plots on a yearly basis. Then the nitrogen conversion factor is given by the ratio of emitted soil  $N_2O-N$  to the fertilizer nitrogen applied. The  $N_2O$  conversion factor induced by N fertilizer was calculated using

$$\text{EF applied N}\left(\%\right) = 100 \times \frac{\left(N_2 \text{O-N}_{\text{fert}} - N_2 \text{O-N}_{\text{non-fert}}\right)}{N_{\text{fert}}}.$$

In the next step, emission factors are expressed as CO<sub>2</sub> equivalents per unit mass of fertilizer nitrogen in line with international greenhouse gas accounting practice [22].

# 2.5. Indirect N<sub>2</sub>O emissions

Two indirect emission paths of N<sub>2</sub>O after application of N mineral fertilizer were involved in this study-the volatilization of NH3 and NO<sub>x</sub> and the leaching of NO<sub>3</sub><sup>-</sup>. It is suggested that both emission paths subsequently yield N<sub>2</sub>O emissions. For the first path, default values were applied according to IPCC [22]. The second path, the leaching of NO<sub>3</sub><sup>-</sup> from soil, was determined during one complete year of the study period using the resin core technique [26]. Resin cores had an inner diameter of 55 mm and a length of 100 mm and were filled with a mixture of pre-washed Amberlite MB 20 and quartz sand. Resin cores with 8 replicates at each plot were installed on 28 October 2005. Mounting was undertaken laterally from the side of a soil pit 90 cm beneath the soil surface. One year later on 7 November 2006, the cores were collected and cut into 3 layers (0–20, 20-60, 60-100 mm). The total fresh weight of each fraction was determined and an aliquot of about 10 g was extracted with 100 mL of 1 M KCl. NO<sub>3</sub><sup>-</sup> contents were determined photometrically with an automatic flow analyzer (RFA-300, Alpkem Corp., Clackamas, OR and Skalar, The Netherlands). The top resin layer had the highest N concentrations and the middle one the lowest N concentrations, verifying that the capacity of the resin was sufficient to prevent break-through and no nutrients were lost by leaching. Only the top two resin layers were taken for flux calculations as described by Lehmann et al. [27]. Fertilizer-related leached nitrate, which accumulated in the resin cores, was calculated as

$$NO_3\text{-}N_{leach} = (NO_3\text{-}N_{leach/fert} - NO_3\text{-}N_{leach/non\text{-}fert}).$$

According to the guidelines of IPCC [22], indirect N<sub>2</sub>O emissions are obtained by applying a conversion factor of 0.75%. The emission

factor for  $N_2O$  followed by  $NO_3^-$  leaching was calculated using the expression

$$EF = 0.75\% \times \frac{44}{28} \times \frac{NO_3\text{-}N_{leach}}{N_{fert}} \,. \label{eq:effective}$$

Finally, CO<sub>2</sub> equivalents are obtained from multiplying the emission factor with the global warming potential of 298.

# 2.6. Harvesting of annual and perennial energy crops

Poplar was harvested during the study period in January 2002 and December 2005 by contrast with winter rye, which was harvested six times during the study period in the months July and August of 1999, 2001, 2003, 2005, 2006 and 2007. In the intervening periods, this site was cultivated with hemp (2000), rape (2002), and triticale (2004).

Five subplots were harvested from each plot. Within the poplar plots, harvesting was performed from  $10\,\mathrm{m}^2$  subplots with a tractor-mounted cutter chipper. In the case of rye, biomass subplots were mowed from an area of  $4\,\mathrm{m}^2$ . Biomass was weighed using an electronic balance in the field. Oven-dry mass was obtained from an aliquot of chopped biomass after drying at  $106\,^{\circ}\mathrm{C}$ .

# 2.7. Statistical analyses

Data of the factorial experiment were subjected to a one-way analysis of variance. Student's t-test was used for comparisons of results. Where significant effects were found, means were compared using the least significant difference test (LSD) at p < 0.05. All statistical analyses were performed using SAS-JMP 8.0.

# 3. Results

# 3.1. Biomass yield

During the study period of nine years, the plots with poplar were harvested twice during winter months in 2001/2002 and 2005/2006. Biomass yields from the shoots of rye were derived from annual harvesting, which was followed by ploughing and sowing in contrast to poplar, which remained alive and grew again after harvesting.

At both harvests of poplar there was a decrease in biomass yield due to fertilization, which however could not be proved statistically (2001/2002: p = 0.066, 2005/2006: p = 0.303). The average was  $10.18 \text{ t ha}^{-1} \text{ yr}^{-1}$  at the non-fertilized control plot and  $9.08 \text{ t ha}^{-1} \text{ yr}^{-1}$  at the plot fertilized with  $150 \text{ kg ha}^{-1} \text{ yr}^{-1}$  (Table 2).

In case of the annual rye crop, significant fertilization effects were obtained. The mean biomass yield of fertilized rye (75 kg N, 150 kg N) exceeded that of the non-fertilized site by 38% and 47%, respectively. Very low yields were obtained for rye in the dry years of 1999 and 2003, when precipitation was below 450 mm yr $^{-1}$ . For the biomass of poplar and rye, carbon contents of 48% and 50% respectively were taken from Lamlom and Savidge [28] and from own measurements as a basis for further calculation of  $\rm CO_2$  equivalents.

# 3.2. Soil N<sub>2</sub>O emissions

Seasonal variability of  $N_2O$  emissions has already been reported earlier for the experimental field under study [29–31]. In Fig. 2, only two years from the nine-year study period are presented as an example, showing the seasonal pattern with  $N_2O$  emission rates, which are generally higher during the warm season in contrast to the cold season. Even the non-fertilized plots show this pattern, reflecting the effect of temperature-dependent mineralization of

**Table 2** Above-ground biomass yield of energy crops at the ATB experimental field. Poplar was harvested every four years in contrast to rye harvested annually. All yield data are expressed on a yearly basis in tons dry matter  $ha^{-1}$  yr<sup>-1</sup>. Values in one line followed by the same letter are not significantly different at p < 0.05.

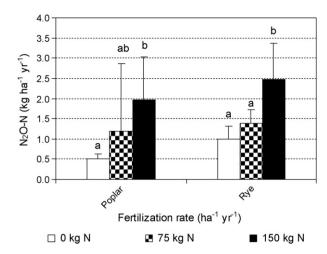
		0 kg N	75 kg N	150 kg N
Poplar	2001/2002	9.91 a	9.62 a	9.07 a
	2005/2006	10.45 a	9.31 a	9.09 a
	Mean	10.18 a	9.46 ab	9.08 ab
Rye	1999	4.41 a	5.46 a	5.69 a
	2001	5.99 a	9.34 b	9.26 b
	2003	3.26 a	5.98 ab	6.37 b
	2005	4.52 a	7.78 b	9.54 c
	2006	8.33 a	10.34 b	9.39 c
	2007	5.14 a	4.96 a	6.34 b
	Mean	5.28 a	7.31 b	7.77 b

organic nitrogen compounds in the soil. On most days,  $N_2O$  emission rates followed the intensity of N fertilization, particularly after the periods of fertilization.

One extraordinary case was observed in 2006, when there were long-lasting high emission rates up to 1160  $\mu g \ N_2 O \ m^{-2} \ h^{-1}$  in the poplar plot fertilized with 75 kg N ha $^{-1}$  (Fig. 2). It is suggested that the nitrogen excess was promoted after harvest and fertilizer application in April 2006. Thus a considerable amount of nitrogen could have been released as  $N_2O$ .

During the nine-year study period, the mean  $N_2O$  emission rate in the non-fertilized plot of rye was  $1.0 \text{ kg } N_2O-N \text{ ha}^{-1} \text{ yr}^{-1}$  and thus twice as high as in non-fertilized poplar (Fig. 3). This background emission is relatively high by comparison with studies from Malhi and Lemke [32], who found values of up to  $0.3 \text{ kg } N_2O-N \text{ ha}^{-1} \text{ yr}^{-1}$ . Even lower values of  $0.14 \text{ kg } N_2O-N \text{ ha}^{-1} \text{ yr}^{-1}$  were reported by Groeningen van et al. [33]. Significant differences in  $N_2O$  emission rates were obtained between the control and the fertilization rate of  $150 \text{ N ha}^{-1} \text{ yr}^{-1}$ .

Considering  $N_2O$  emissions on our experimental field during the study period, the fertilizer-induced conversion factors were



**Fig. 3.** Mean  $N_2O$  fluxes (+1SD) at the ATB experimental field between 1999 and 2007 (poplar n = 9, rye n = 6). Different letters between two bars reflect significant difference of p < 0.05%.

0.75% for rye and 0.95% for poplar each as a mean of the 75 kg N and 150 kg N fertilization rate, if the  $N_2O$  emission rates of nonfertilized plots were taken into account. These factors are higher than those measured on the same experimental field earlier [29] and those measured at another site on sandy soils in the Netherlands [33].

# 3.3. NH<sub>3</sub> and NO<sub>x</sub> volatilization

According to IPCC [22], it can be assumed that 10% of applied nitrogen mineral fertilizer is emitted by volatilization in the form of NH<sub>3</sub> and NO<sub>x</sub>. IPCC [22] has set a default value of 1% of the volatilized fraction, which forms N<sub>2</sub>O after deposition and transformation. This kind of emission is considered as indirect N<sub>2</sub>O emission corresponding to 0.47 kg CO<sub>2</sub> per kg nitrogen in CAN (Table 3).

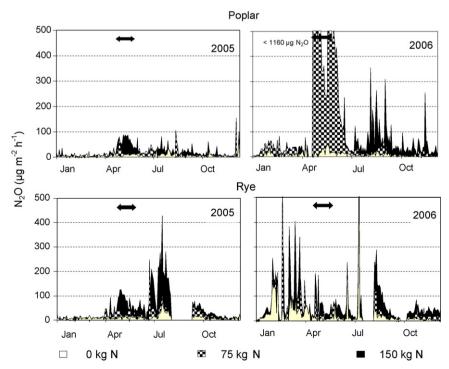


Fig. 2. Daily N<sub>2</sub>O emission from perennial and annual crop plots at the ATB experimental field during a period of two years. Arrows indicate the period of CAN application.

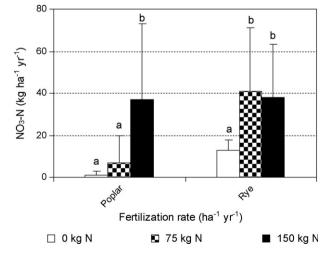
**Table 3**Greenhouse gas emissions from production and application of 1 kg nitrogen in CAN fertilizer. CO<sub>2</sub> equivalents derive from a global warming potential of 298:1 for N<sub>2</sub>O:CO<sub>2</sub>. Emission factors (EF) are expressed as unit mass per fertilizer product.

Source of greenhouse gas	CO <sub>2</sub> equivalents	Reference
Direct fertilizer-induced $\rm N_2O$ emission from soil EF=0.75% for rye EF=0.95% for poplar	$3.51\mathrm{kgCO_2}$ $4.45\mathrm{kgCO_2}$	This study This study
Indirect $N_2O$ emission from leaching EF=0.75% × 0.272 (leached fraction)=0.204% for rye EF=0.75% × 0.157 (leached fraction)=0.118% for poplar	$0.96kgCO_2$ $0.55kgCO_2$	This study This study
Indirect $N_2O$ emission from $NH_3$ and $NO_x$ volatilization and redeposition EF = $1\%\times0.1$ (volatilized fraction) = $0.1\%$	$0.47\mathrm{kg}\mathrm{CO}_2$	IPCC [22]
$\text{CO}_2$ emission from energy consumption required for NH $_3$ synthesis EF = 2.104 kg $\text{CO}_2$ (kg NH $_3$ ) $^{-1}$	2.61 kg CO <sub>2</sub>	IPCC [22]
$N_2O$ emission from $HNO_3$ manufacturing EF = $5.5$ kg $N_2O$ (Mg $HNO_3$ ) <sup>-1</sup>	$3.67\mathrm{kgCO_2}$	UBA [34]
CO <sub>2</sub> emissions from CaCO <sub>3</sub> manufacturing and decaying after soil application	$0.45\mathrm{kg}\mathrm{CO}_2$	IPCC [22]

# 3.4. Leaching

During a complete year from 2005 to 2006, up to 41 kg nitrogen was leached as nitrate (Fig. 4). Generally, the higher the fertilization rate, the higher the amount of leached nitrate. A survey on the experimental field had shown that the different fertilization rates were clearly reflected by the mean soil NO<sub>3</sub>–N with 0.5 mg kg $^{-1}$ , 0.9 mg kg $^{-1}$  and 1.6 mg kg $^{-1}$  for the nonfertilized plots and the plots fertilized with 75 kg N and 150 kg N, respectively [31]. Taking a soil density of 1.5 g cm $^{-3}$  [5], that resulted in 2.3 kg N ha $^{-1}$ , 4.0 kg N ha $^{-1}$  and 7.2 kg N ha $^{-1}$  within the 0–30 cm horizon.

The loss of fertilizer nitrogen adjusted by the loss from the nonfertilized plot was 8% and 24% for poplar and 37% and 17% for rye in terms of fertilizer rates of 75 and 150 kg N. These leaching ratios are in the range of 30% given for regions with a positive water balance by the IPCC [22]. Leached nitrogen compounds may result in indirect N<sub>2</sub>O emissions and were calculated to be 0.118% and 0.204% of applied N for poplar and rye, respectively (Table 3). Transforming the indirect N<sub>2</sub>O emissions to CO<sub>2</sub> equivalents resulted in values ranging between 0.26 kg CO<sub>2</sub> (kg N)<sup>-1</sup> and 0.84 kg CO<sub>2</sub> (kg N)<sup>-1</sup> for poplar and between 0.60 kg CO<sub>2</sub> (kg N)<sup>-1</sup> and 1.32 kg CO<sub>2</sub> (kg N)<sup>-1</sup> for rye, respectively. The mean value for each energy crop is given in Table 3.



**Fig. 4.** Amount of NO<sub>3</sub>–N (mean + 1SD), which was leached and fixed at a depth of 0.9 m at the ATB experimental field during a period of one year from 2005 to 2006 (n = 8). Different letters between two bars reflect significant difference of p < 0.05%.

# 3.5. $CO_2$ equivalents from the manufacturing of calcium ammonium nitrate (CAN)

CAN fertilizer is a mixture of NH<sub>4</sub>NO<sub>3</sub> and CaCO<sub>3</sub> with a mass N:CaCO<sub>3</sub> ratio of about 1:1, containing up to 27% nitrogen and 10% calcium. Nitrogen occurs half in the nitrate form and half in the ammoniacal form. One kilogram of nitrogen from CAN fertilizer corresponds to 2.846 kg NH<sub>4</sub>NO<sub>3</sub>, which is produced from NH<sub>3</sub> and HNO<sub>3</sub>. The majority of greenhouse gas emissions are made up of CO<sub>2</sub> emissions from ammonia synthesis and N<sub>2</sub>O emissions from nitric acid production. Considering an emission factor of 2.104 kg CO<sub>2</sub> for the synthesis of 1 kg NH<sub>3</sub> [22] and the global warming potential of 298:1 for N<sub>2</sub>O:CO<sub>2</sub>, we obtain a CO<sub>2</sub> equivalent of 2.61 kg, which corresponds to the production of 1 kg CAN nitrogen (Table 3). Furthermore, when HNO<sub>3</sub> is manufactured by catalytic NH<sub>3</sub> oxidation, N<sub>2</sub>O is released as a by-product. The amount of N<sub>2</sub>O depends on the state of technology and is subject to a high variability between  $2 \text{ kg N}_2\text{O} \text{ (Mg HNO}_3)^{-1}$  and  $9 \text{ kg N}_2\text{O} \text{ (Mg HNO}_3)^{-1}$  [22]. Applying an emission factor of 5.5 kg  $N_2O$  (Mg  $HNO_3$ )<sup>-1</sup> [34], the production of required 2.241 kg HNO<sub>3</sub> yields CO<sub>2</sub> equivalents of 3.67 kg CO<sub>2</sub>. In the case of CaCO<sub>3</sub> production only about 0.011 kg CO<sub>2</sub>  $(kg CaCO_3)^{-1}$  is emitted [22]. This is far below the emission of  $0.44 \text{ kg CO}_2 (\text{kg CaCO}_3)^{-1}$  which results from application to soil. The overall emissions released to produce 1 kg N as CAN fertilizer can be expressed as 6.29 kg CO<sub>2</sub> equivalents.

# 3.6. $CO_2$ balance of the production of energy crops

The total gain of  $CO_2$  equivalents, which may be calculated from the carbon fixed in the biomass, ranged between 64.0 t  $CO_2$  ha<sup>-1</sup> and 71.6 t  $CO_2$  ha<sup>-1</sup> during a rotation of four years for poplar, that are  $16.0 \text{ t } CO_2$  ha<sup>-1</sup> yr<sup>-1</sup> and  $17.9 \text{ t } CO_2$  ha<sup>-1</sup> yr<sup>-1</sup> and between  $9.7 \text{ t } CO_2$  ha<sup>-1</sup> yr<sup>-1</sup> and  $14.2 \text{ t } CO_2$  ha<sup>-1</sup> yr<sup>-1</sup> for rye (Fig. 5). If greenhouse gas emissions from the soil and from the production of mineral fertilizer (CAN) are included, then  $CO_2$  fixation is counteracted by the intensity of fertilization. The fertilization rate of  $150 \text{ kg ha}^{-1} \text{ yr}^{-1}$  resulted in a reduction of  $2.1 \text{ t } CO_2 \text{ ha}^{-1} \text{ yr}^{-1}$  and  $2.3 \text{ t } CO_2 \text{ ha}^{-1} \text{ yr}^{-1}$  for poplar and rye, respectively.

Among direct and indirect greenhouse gas emissions, direct  $N_2O$  soil emissions contributed the major amount of  $CO_2$  equivalents. The maximum rate of indirect  $N_2O$  emissions as a consecutive process of  $NO_3$  leaching and  $NH_3$  and  $NO_x$  volatilization was 0.20 t  $CO_2$  ha<sup>-1</sup> yr<sup>-1</sup> in the case of poplar (150 kg N) and thus did not play an important role. Considering all calculated and estimated  $CO_2$  equivalents released by the production of poplar and rye, the positive effect of  $CO_2$  saving is reduced by 1.3% (poplar

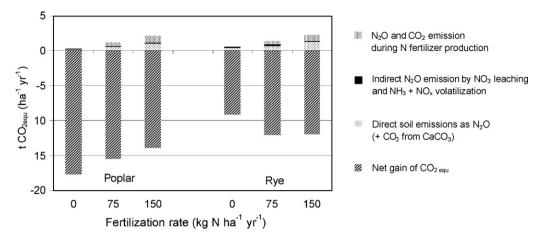


Fig. 5. CO<sub>2</sub> balance during the production of energy crops at the ATB experimental field between 1999 and 2007.

0~kg~N) and 15.7% (rye 150~kg~N). There is a net gain of  $CO_2$  equivalents between  $13.9~t~CO_2~ha^{-1}~yr^{-1}$  and  $17.7~t~CO_2~ha^{-1}~yr^{-1}$  for poplar and between  $9.2~t~CO_2~ha^{-1}~yr^{-1}$  and  $12.1~t~CO_2~ha^{-1}~yr^{-1}$  for rye. From 1999 to 2007 the reduction of global warming equivalents (GWE) by greenhouse gas emissions was highest at the non-fertilized plot of poplar and at the fertilized plots of rye with no difference between the two fertilization rates (Fig. 5).

#### 4. Discussion

#### 4.1. Mineral N fertilizer

The production of N fertilizer contributes to a high share of fossil energy consumption in crop production, which has to be taken into account when criteria for sustainable agricultural production are defined. Furthermore, even at the pre-farm stage the production of N fertilizer leads to considerable N<sub>2</sub>O emissions, which reduce the positive effect of energy crops. At the on-farm stage, N<sub>2</sub>O emission occurs as a natural process, which is accelerated after application of N fertilizer. Mean values of N<sub>2</sub>O emission in this study are still in the range of 1.0%, which is generally considered as a tolerable mean conversion factor for agricultural soils of temperate regions, used as default value for direct N2O emissions in greenhouse gas inventories [22]. When considering the non-fertilized poplar plots, the conversion of 0.95% of fertilizer nitrogen to N2O-N appears to be rather high for a perennial crop. It is mainly caused by the high N<sub>2</sub>O emission rates in 2006, pointing to the strong impact of hot spot events on the nitrogen conversion factor [4]. Even higher fertilizer-induced N2O conversion factors of up to 5% were reported by Rudaz et al. [10] on a permanent pasture with a sandy loam, and by Flechard et al. [35] on grassland soils in Europe. Also high N<sub>2</sub>O conversion factors were obtained from Australian sugarcane soils [36]. The values varied between 1.2 and 6.7%, indicating higher and more variable N<sub>2</sub>O emissions as given in the IPCC guidelines. Up-scaling of N fluxes remains highly uncertain as long as no better data exist for regional-specific conditions and for agricultural practices. Therefore, N flux verification procedures are needed to assess compliance with international protocols [37].

If CO<sub>2</sub> equivalents from direct and indirect greenhouse gas emissions as a consequence of fertilization are added, then up to 16% of the CO<sub>2</sub> fixed in the biomass is counterbalanced in this study. This is in accordance with data from Lewandowski and Schmidt [38], who found that 5–20% of the amount of energy contained in the harvested biomass is necessary as input to obtain the biomass being produced for solid fuel. Therefore those crops that need little or no N fertilization do well for the environment.

# 4.2. Energy and C balance

Energy input is needed for biomass production and the net energy yield provides a picture of the efficiency of the energy crop in replacing fossil energy and the avoidance of greenhouse gas emissions.

Cumulative energy demand for the production of solid biofuels can be distinguished between cereals and poplar with about 20 GJ  $ha^{-1}$   $yr^{-1}$  and 3–7 GJ  $ha^{-1}$   $yr^{-1}$ , respectively [39]. The major role in producing and harvesting energy crops is played by the energy input needed for cultivation as shown for poplar by Gasol et al. [40]. This includes the use of fertilizer, which was less effective in poplar compared with rye in our study. If poplar is produced and N fertilizer can be saved, the final cost must be lower for poplar than for energy rye. Besides the energy use in fertilizer production and field N<sub>2</sub>O emissions, the carbon content of the soil may also play a key role in the net global warming potential. Mosier et al. [41] reported that soil organic carbon storage decides whether a system is a sink or a source of CO<sub>2</sub>. The carbon tradeoff on the experimental field at Potsdam has been studied recently by Strähle [42], who calculated that the soil organic carbon stocks in the top 30 cm of perennial crops were significantly increased compared with annual crops. More long-term studies are needed to verify these first results, which are based on soil conditions with an unusually rich A-horizon. This could have influenced the tree growth and also the organic carbon storage in the soil.

The  $\rm CO_2$  balance presented here derives from only a few selected elements of a LCA, which are believed to play important roles. Post-harvest processes as well as additional greenhouse gas emitted during the combustion of biomass are reportedly also relevant [6], but were outside our system boundary.

# 4.3. Assessment of energy crops

As far as possible, each nitrogen pathway was validated by experimental data, which should reduce the uncertainty in assessing the production of specific energy crops. On the basis of the CO<sub>2</sub> balance deriving from our nine-year study, perennial crops obviously have some advantages over annual crops as already reported by Kaltschmitt et al. [43], Hanegraaf et al. [7] and Rowe et al. [44]. An overall environmental assessment of biomass production has to include other features besides the input of fertilizers, such as for example the input of pesticides. In our study however, we waived the use of pesticides, which are commonly used primarily for annual crops. Another economic advantage of perennial crops is a harvest cycle of four or more years, which allows a low frequency of required machine operations. This

generally results in ecological benefits of reduced N<sub>2</sub>O emissions due to lower soil disturbance.

The real environmental and economic benefit of energy crops depends on region-specific conditions as well as on the system boundary chosen for the assessment. A complete LCA as recently demonstrated by Biswas et al. [14] for wheat production in Australia has clearly distinguished between greenhouse gas emissions from pre-farm, on-farm and post-farm stages. In that case, pre-farm processes such as the production of mineral fertilizer and herbicides played the major role in greenhouse gas emissions. It could also be shown that in situ measurements of N<sub>2</sub>O emissions from the soil have a considerable effect on the final result of a LCA, although the N<sub>2</sub>O soil emission factor, determined in the Australian study, was only one sixth of the IPCC default value of 1%. This underlines once more the great value of data from in situ measurements, which should be consulted for the careful assessment of energy crops as far as available. Crutzen et al. [6] concluded from the evaluation of global N<sub>2</sub>O budgets, taking into account global N fixing (biologically, by industrial fixation and by fossil fuel burning), that the overall N conversion factor should be in the range between 3% and 5%. Based on this, they derived a criterion for the ratio of N to dry matter in the plant material, which indicates to what degree the reduced global warming achieved by using biofuels instead of fossil fuel as energy sources is counteracted by release of N<sub>2</sub>O. This discrepancy between a "global N2O emission factor" related to "global N fixing" and the IPCC default value for N fertilization results from the different ways of consideration. IPCC factors are given for detailed processes, e.g. fertilization or volatilization. Livestock production cycles N via the nath:

forage  $\rightarrow$  feeding animals  $\rightarrow$  manure storage

 $\rightarrow$  manure application as fertilizer  $\rightarrow$  plants  $\rightarrow$  forage.

N<sub>2</sub>O is emitted directly due to fertilization, but also indirectly via volatilization and leaching in the course of each cycle. When the total N<sub>2</sub>O emissions of the whole cycle of livestock breeding are related to the total input of fixed N (mainly mineral fertilizer), then emission factors will result that are several times higher compared with the N<sub>2</sub>O emission factor of annually applied N fertilization. The best way of producing energy crops with a high nitrogen and energy use efficiency is to choose a crop, which efficiently transforms the growth factors into harvestable energy. As outlined by Lewandowski and Schmidt [38], this is realized by the production of perennial  $C_4$  grasses such as Miscanthus  $\times$  giganteus. giganteus. On cooler sites, the production of short rotation coppices is suggested in order to use nutrient and energy resources most efficiently. The use of former arable soils particularly is recommended, since most agricultural soils are enriched with macronutrients such as nitrogen [45]. After all, the experimental field under study had been a fruit-growing plantation with input of organic residues before the experimental field was established. This should have enabled a good and in part excessive nutrient supply in the soil. Additional hot spots of N<sub>2</sub>O emission, which could not be explained in detail, could have been the reason for the relatively high conversion of mineral N to N<sub>2</sub>O.

It could be shown that on the loamy sand under study, poplar had a greater potential for reducing  $CO_2$  emissions than rye. In contrast to rye with its highest mitigation of  $CO_2$  equivalents at the fertilization rates of 75 kg N ha<sup>-1</sup> yr<sup>-1</sup> and 150 kg N ha<sup>-1</sup> yr<sup>-1</sup>, an inverse pattern occurred for poplar. Thus the saving of nitrogen fertilizer for biomass production characterizes poplar as a valuable energy crop.

# 5. Conclusions

The production of energy crops facilitates the removal of CO<sub>2</sub> and thermal conversion of these crops is one way of substituting

them for fossil fuels. However, a distinction must be made between different crops. How far their production can be considered as economical and environmentally acceptable depends on the intensity of negative drawbacks, such as the emission of  $N_2O$ . One of the most important factors controlling  $N_2O$  emissions is the application of nitrogen fertilizer. In this study, there was a loss of  $CO_2$  equivalents fixed in the biomass of up to 16%. Production and application of N fertilizer is the main reason for direct and indirect greenhouse gas emissions. Consequently, reducing the input of N fertilizers to the fields would be one way of mitigating  $N_2O$  emissions and thus enhancing the  $CO_2$  mitigation potential of energy crops. Therefore further research should focus on energy crops and new genotypes, which have both a high biomass yield and a low nitrogen requirement.

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